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REGIO- AND STEREOSELECTIVE SYNTHESIS OF *gem*-DIFLUOROCYCLOPROPANES USING 4-BROMO-4,4-DIFLUOROCROTONATE

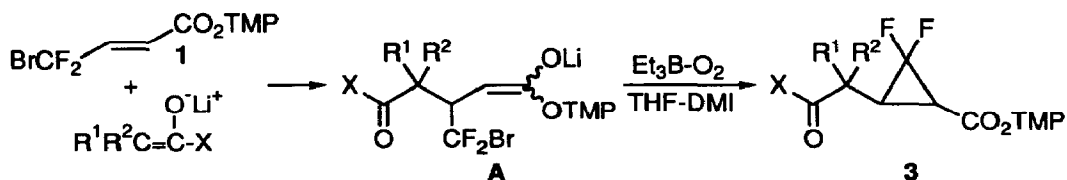
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Abstract: *Regio- and stereoselective formation of the functionalized gem-difluorocyclopropane derivative was achieved through the Michael addition of lithium enolate of ester or amide with 4-bromo-4,4-difluorocrotonate 1 followed by the triethylborane-O₂ mediated intramolecular substitution reaction.*

Cyclopropane derivatives have been recognized as an important class of compounds found in natural and unnatural substances, with particularly significant application in bio- and medicinal chemistry.^{1,2} For example, conformationally restricted analogs of glutamic acid having the cyclopropane moiety were studied so as to elucidate the conformations (extended and folded forms) necessary for the receptor subtype specificity,³ and substrates having the cyclopropane moiety were found to be mechanism dependent inhibitors of certain enzymes.^{1a,4} For such chemically modified biologically active compounds, the introduction of fluorine atom(s) onto cyclopropane ring would lead to interesting results in consideration of characteristic features of fluorinated compounds.^{5,6} It should thus be needed to develop an efficient method for the preparation of suitably functionalized fluorinated cyclopropanes in a stereoselective manner. In this paper, we report a regio- and stereoselective synthesis of functionalized *gem*-difluorocyclopropane **3** through the sequential Michael addition of lithium enolate of ester or amide with 2,4,6-trimethylphenyl (TMP) ester of 4-bromo-4,4-difluorocrotonate **1**^{7,8} and the triethylborane mediated intramolecular substitution reaction (Scheme 1).⁹



Scheme 1

At first, we examined the reaction conditions for the cyclopropanation reaction of **1** with *t*-butyl acetate **2a**. Results are shown in Table 1. Reaction of **1** with lithium enolate of **2a** proceeded at low temperature within a short period to give the Michael addition product **4a**, but the formation of the difluorocyclopropane **3a** was not observed (entry 1). Elongation of reaction period or addition of a polar solvent such as DMI (1,3-dimethyl-2-imidazolidinone) produced a complex mixture (entry 2). These results indicate that activation of the CF₂-Br bond would thus appear essential following the Michael addition of the enolate to **1**. After several attempts, we found that triethylborane(Et₃B)-O₂ and a polar solvent such as DMI were effective additives for this reaction. Thus, after treating **1** with lithium enolate of **2a** in THF at -78°C for 15 min, Et₃B (3 equiv.), O₂ (5 ml bubbled via syringe) and DMI were added and the reaction mixture was stirred at -20°C for 3 h to give the cyclopropane **3a** in 71% yield along with a small amount of the Michael addition product **4a** (entry 5). **3a** was obtained as a single compound and its *trans* stereochemistry was confirmed from the NMR spectrum (*J*₂₋₃=7.3 Hz)¹⁰ and the regioselectivity in the reaction was assessed by functional selective conversion of *t*-butyl ester group to the alcohol.¹¹

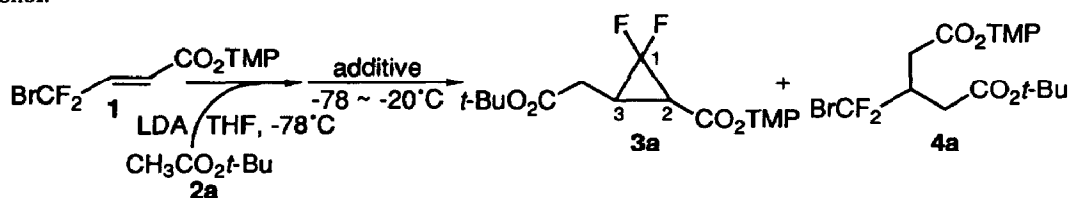


Table 1. Reaction of **1** with *t*-Butyl Acetate **2a**^a

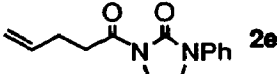
entry	Et ₃ B (eq)	DMI ^b	yield of 3a (%)	yield of 4a (%)
1	none	none	0	70
2	none	10 vol%	0	— ^c
3	0.5	none	26	32
4	3.0	none	17	47
5	3.0	10 vol%	71	7

^a Reaction was carried out for 15 min at -78°C without the additive(s), then for 3 h at ca -20°C after addition of additive(s). ^b DMI=1,3-Dimethyl-2-imidazolidinone. ^c Complex mixture. **4a** was detected by TLC and ¹⁹F-NMR spectrum of the reaction mixture.

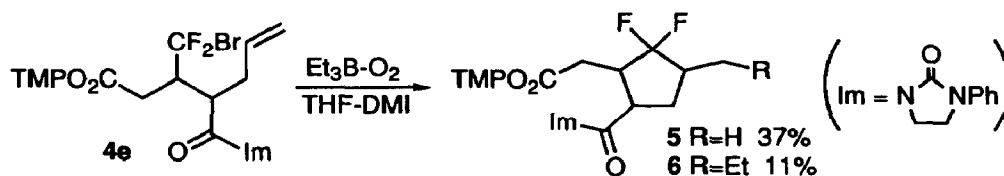
In a similar manner, ester and amide enolates reacted with **1** in the presence of Et₃B-O₂ to give the corresponding difluorocyclopropane **3** (Table 2). A complete regio- and stereoselectivity in the formation of the cyclopropane **d**₂-**3b** was observed for the deuterated acetate **d**₃-**2b** (entry 2). With the TMP ester of α,α -disubstituted carboxylic acid **2c**, the *trans*-cyclopropanes **3c** was obtained in a good yield (entry 3).

It has been reported that trialkylborane reacts with oxygen to generate an alkyl radical at low temperature and Et₃B-O₂ system is an effective radical initiator for tinhydride or alkyl iodide and bromide.^{12,13} In the present cyclopropane formation, Et₃B-O₂ may possibly initiate the radical cleavage of CF₂-Br bond of the intermediate Michael adduct (**A** in Scheme 1). To examine the cleavage of CF₂-Br bond by Et₃B-O₂ system, a control experiment was conducted by using the Michael adduct **4e**, which possesses a double bond as a radical acceptor. Thus, by treating the Michael adduct **4e** with Et₃B-O₂ in THF-DMI at -20°C for 3 h, the 5-exo

Table 2. Et₃B-O₂ Mediated Cyclopropanation Reaction^a

entry	2	3 (yield, %) ^b	4 ^c (yield, %) ^b
1	CH ₃ CO ₂ TMP 2b	3b (47)	4b (18)
2	CD ₃ CO ₂ TMP d ₃ -2b	d ₂ -3b (48)	d ₂ -4b (17)
3	(CH ₃) ₂ CHCO ₂ TMP 2c	3c (73)	4c (8)
4	CH ₃ CON(CH ₃) ₂ 2d	3d (51)	4d (23)
5	 2e	3e (75)	4e (3) ^d

^a Reaction conditions; THF, -78°C, 15 min for Michael addition, then Et₃B, O₂ and DMI, ca -20°C, 3h. ^b Isolated yield. ^c Michael addition product. ^d The cyclopentane derivative 5 (see Scheme 2) was also isolated in 2.5% yield.



Scheme 2

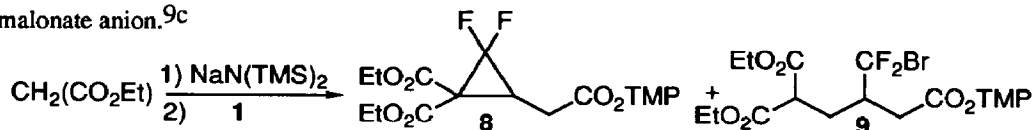
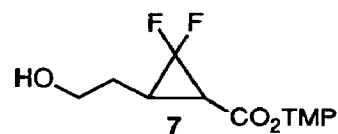
cyclized products 5 and 6 were obtained in 37% and 11% yield, respectively, along with the recovery of 4e in 35% (Scheme 2). Under the sequential Michael addition and cyclopropanation conditions, the *N*-(4-pentenyl)imidazolidinone 2e afforded the cyclopropane 3e in 75% yield along with the 5-*exo* cyclized product 5 (2.5%) and the Michael adduct 4e (3.4%), respectively (Table 2, entry 5). These results may suggest that the formation of the cyclopropane 3e from the intermediate A is much faster than the 5-*exo* cyclization reaction. Investigation of the mechanistic aspects of the present reaction in detail is a future subject.

In conclusion, we have shown that bromodifluorocrotonate 1 is an efficient building block¹⁴ for the preparation of functionalized difluorocyclopropanes through the activation of CF₂-Br bond by Et₃B-O₂ in an intramolecular substitution reaction with lithium enolate.

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14. Bromodifluorocrotonate 1 showed a high reactivity as a Michael acceptor with active methylene compounds. Reaction of 1 with malonate anion provided the Michael adduct 9 and/or the cyclopropane 8 depending on the reaction conditions: (a) 9, 71% (THF, rt); (b) 8, 64% (THF, reflux, 2 h); (c) 8, 66% (THF-DMI, rt, 24 h). In the case of 4-bromocrotonate, only direct S_N2 displacement occurred with malonate anion.^{9c}



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